Catalysis of GTP Hydrolysis by Small GTPases at Atomic Detail by Integration of X-ray Crystallography, Experimental, and Theoretical IR Spectroscopy

Received for publication, February 25, 2015, and in revised form, July 17, 2015. Published, JBC Papers in Press, August 13, 2015, DOI 10.1074/jbc.M115.648071

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Background: The magnesium position in the active site of GTPases varies among x-ray structures of Ran and related GTPases.

Results: The magnesium in both Ras and Ran is always coordinated to the β- and γ-phosphate.

Conclusion: Other factors like steric hindrance by the Tyr-39 side chain influence the hydrolysis rate in Ran.

Significance: The new information allows an improved view on the catalytic mechanism of GTPases.

Small GTPases regulate key processes in cells. Malfunction of their GTPase reaction by mutations is involved in severe diseases. Here, we compare the GTPase reaction of the slower hydrolyzing GTPase Ran with Ras. By combination of time-resolved FTIR difference spectroscopy and QM/MM simulations we elucidate that the Mg2+ coordination by the phosphate groups, which varies largely among the x-ray structures, is the same for Ran and Ras. A new x-ray structure of a Ran-RanBD1 complex with improved resolution confirmed this finding and revealed a general problem with the refinement of Mg2+ in GTPases. The Mg2+ coordination is not responsible for the much slower GTPase reaction of Ran. Instead, the location of the Tyr-39 side chain of Ran between the γ-phosphate and Gln-69 prevents the optimal positioning of the attacking water molecule by the Gln-69 relative to the γ-phosphate. This is confirmed in the RanY39A-RanBD1 crystal structure. The QM/MM simulations provide IR spectra of the catalytic center, which agree very nicely with the experimental ones. The combination of both methods can correlate spectra with structure at atomic detail. For example the FTIR difference spectra of RasA18T and RanT25A mutants show that spectral differences are mainly due to the hydrogen bond of Thr-25 to the α-phosphate in Ran. By integration of x-ray structure analysis, experimental, and theoretical IR spectroscopy the catalytic center of the x-ray structural models are further refined to sub-Å resolution, allowing an improved understanding of catalysis.

Small GTPases are molecular switches, which regulate many cellular processes (1, 2). The “on” state is characterized by a specific conformation of the switch region of the proteins, which is induced by GTP binding, whereas the “off” state is obtained with GDP in the binding pocket. The “switch off” is performed by GTP hydrolysis, which can be further accelerated by GTPase activating proteins (3). The “switch on” is facilitated by nucleotide exchange, usually effected by guanine nucleotide exchange factors. The small GTPase Ras is the central regulator in cell growth and differentiation (4). Ras is mutated in about 25% of human tumors. The related small GTPase Ran regulates nuclear transport (5, 6) and is involved in spindle formation and nuclear envelope formation of dividing cells (7). The GTP form of Ran is stabilized by Ran-binding proteins (RanBP1 or the structurally similar RanBD domains of the RanBP2 protein) (8).

The reaction rate of the intrinsic hydrolysis reaction among GTPases varies. Especially Ran has a very slow reaction rate compared with other GTPases. This is important to maintain the Ran gradient: Ran-GTP should be concentrated mainly in the nucleoplasm and Ran-GDP in the cytoplasm (9). Fast intrinsic GTP hydrolysis by Ran in the nucleoplasm (i.e. without GTPase activating protein) would decrease the gradient and impact the nuclear transport, which can lead to severe diseases like the Hutchinson-Gilford Progeria Syndrome (10). Indeed, GTP hydrolysis is roughly 26 times slower in Ran and about 13 times slower in Ran-RanBD1 than in Ras (11). Here, we address the atomic level the mechanism underlying the difference in the Ran and Ras hydrolysis rate.

One potential candidate for the difference is the position of the magnesium ion. FTIR experiments have shown Mg2+ to be one of the key players for catalyzing GTP hydrolysis due to its ability to shift negative charge within the GTP (12). The corresponding conformational change and charge shift of the triphosphate could be a reason for the different hydrolysis rates. Indeed, the position of the Mg2+ varies among available GTPase crystal structures. Fig. 1 shows a comparison of the Mg2+ coordination in Ras and Ran with bound GTP or GTP analogues based on x-ray structural models. The x-ray struc-
The Mg\textsuperscript{2+} coordination in the x-ray structures of (a) Ras-GTP (PDB code 1QRA, light blue carbon atoms), (b) Ran-Gpp(NH)p-RanBD1 (PDB code 1RRP, light green carbon atoms), and (c) various available Ran x-ray structures with their nucleotides aligned to Gpp(NH)p of PDB code 1RRP (Ran-Gpp(NH)p-RanBD1, Mg\textsuperscript{2+} in magenta) and the Mg\textsuperscript{2+} ions shown as pink spheres. Both, the \(\beta\)- and \(\alpha\)-\(\beta\)-form are found. A similar distribution is found for Ras complexes (Fig. 4). The used x-ray structures in c can be found in supplemental Section S2.

The abbreviations used are: PDB, Protein Data Bank; Gpp(NH)p, guanosine 5'-[(\(\beta\),\(\gamma\)-imido)triphosphate; QM/MM, quantum mechanics/molecular mechanics; cgGTP, caged GTP; NPE-GTP, 1-(2-nitrophenyl)-ethylester of GTP; pHGTP, para-hydroxyphenacyl ester of GTP.

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The Mg\textsuperscript{2+} positions in Ras and Ran x-ray structures. Comparison of the Mg\textsuperscript{2+} coordination in the x-ray structures of (a) Ras-GTP (PDB code 1QRA, light blue carbon atoms), (b) Ran-Gpp(NH)p-RanBD1 (PDB code 1RRP, light green carbon atoms), and (c) various available Ran x-ray structures with their nucleotides aligned to Gpp(NH)p of PDB code 1RRP (Ran-Gpp(NH)p-RanBD1, Mg\textsuperscript{2+} in magenta) and the Mg\textsuperscript{2+} ions shown as pink spheres. Both, the \(\beta\)- and \(\alpha\)-\(\beta\)-form are found. A similar distribution is found for Ras complexes (Fig. 4). The used x-ray structures in c can be found in supplemental Section S2.

Comparison of the kinetics of Ran-RanBD1, RanY39A-RanBD1, Ras, and RasY32A. The normalized absorbance differences of adjacent bands of the GTP and GDP state are shown as a measure for the progress of the hydrolysis. The fits to single exponential functions are shown as continuous lines.

slow hydrolysis rate \((5 \times 10^{-8} \text{ s}^{-1})\) at 303 K \((15)\). The different GTP-Mg\textsuperscript{2+} coordination might explain why the hydrolysis rate of GTP-Mg\textsuperscript{2+} bound to Ran in the RanBP1 complex \((3.5 \times 10^{-5} \text{ s}^{-1} \text{ at 303 K})\) is slower than for GTP-Mg\textsuperscript{2+} bound to H-Ras \((6.8 \times 10^{-4} \text{ s}^{-1} \text{ at 303 K})\) (Fig. 2) \((11)\). FTIR spectroscopy is sensitive to very small changes in the nucleotide environment, and indeed, the spectrum of the GTP-phosphates in Ran is significantly different to GTP in Ras (Fig. 3) \((11)\).

However, available x-ray structures reveal a more complicated picture than described above. In particular, an analysis of the Mg\textsuperscript{2+} position in various GTPase x-ray structures reveals an uncertainty in the Mg\textsuperscript{2+} location. There are also Ran structures available indicating a \(\beta\)-\(\gamma\)-form (Fig. 1c). In contrast, all structures of Ras alone show a \(\beta\)-\(\gamma\)-form (Fig. 4d). However, if Ras complexes are also considered, the position of the Mg\textsuperscript{2+} is again shifting quite systematically (Fig. 4a). The same behavior is also observed for Rab GTPases (Fig. 4, c and e). To investigate this unexpected variability of Mg\textsuperscript{2+} location and obtain additional information about its position in solution, we combine in this study new crystal structures of Ran complexes with improved resolution with theoretical and experimental vibrational spectroscopy.

It has been shown that spectral changes can be decoded into structural changes by the combination of FTIR spectroscopy with biomolecular simulations \((16)\). QM/MM calculations allow the accurate calculation of the vibrational spectrum \((17–21)\). If experimental and theoretical spectra agree, the theoretical structural model is validated and the detailed simulated structural models can be used for interpreting reaction mechanisms and catalysis. Bond length can be calculated with a resolution of 0.01 Å, well below the resolution of x-ray analysis. However, these changes are significant for catalysis. The change of a C-C single bond by 0.1 Å corresponds to 6 kJ/mol, which is about one-quarter of the change in free energy of activation for GTP hydrolysis in water compared with GTP in Ras. A combined QM/MM and FTIR analysis was performed for H-Ras and it was found that a combination of charge shifts, an increase in \(P_{\beta\gamma} P_{\gamma}\) distance, and a strained, staggered conforma-
tion of the GTP induced by the Ras and GAP seem to be responsible for catalysis (16).

Here, we report on an atomic structural model of the triphosphate environment of Ran obtained by the combination of x-ray structure analysis, biomolecular simulations, and FTIR spectroscopy. Our model allows for the elucidation of the correct Mg$^{2+}$ position in solution and permits an investigation of the dramatic difference in Ran and Ras hydrolysis rates.

Materials and Methods

Biomolecular Simulations—Molecular mechanics (MM) simulations were performed with Gromacs 4.0.7 (22, 23), and the hybrid quantum mechanics/molecular mechanics (QM/MM) simulations with Gromacs/Gaussian03 (23–25). The used simulation systems are described in detail in supplemental Table S1. Hydrogen atoms were added to the x-ray structures by the MAXIMOBY algorithm (26). Afterward the protonated structures were checked and amended by the same algorithm. Parameters for the MM and QM/MM simulations were as described previously (12). To calculate theoretical IR features and structural details we used the same workflow as described in an earlier paper (16), but with an enlarged quantum region.

For the QM/MM equilibration run the triphosphate, ribose, Mg$^{2+}$, Ala-18 (Ras)/Thr-25 (Ran), the backbone carboxyl and oxygen atom of Ser-17/Thr-24, a dummy atom between the carboxyl atom, and the $\alpha$ carboxyl atom of Ser-17/Thr-24, the backbone nitrogen atom and its bound hydrogen atom of Leu-19/Phe-26, and the dummy atom between the nitrogen atom and the $\alpha$ carboxyl atom of Leu-19/Phe-26 were treated quantum mechanically. For the first QM/MM energy optimization run the triphosphate, the ribose, the Mg$^{2+}$, Ala-18/Thr-25, the backbone of Ser-17/Thr-24, the backbone of Lys-16/Lys-23, the backbone carboxyl and oxygen atom of Gly-15/Gly-22, a dummy atom between the carboxyl atom and the $\alpha$ carboxyl atom of Gly-15/Gly-22, the backbone nitrogen atom and its bound hydrogen atom of Leu-19/Phe-26, and the dummy atom between the nitrogen atom and the $\alpha$ carboxyl atom of Leu-19/Phe-26 were treated quantum mechanically. For the second QM/MM energy optimization run and the following normal mode analysis the triphosphate, ribose, Mg$^{2+}$, Ala-18/Thr-25, the backbone of Ser-17/Thr-24, the backbone of Lys-16/Lys-23, the backbone carboxyl and oxygen atom of Gly-15/Gly-22, a dummy atom between the carboxyl atom and the $\alpha$ carboxyl atom of Gly-15/Gly-22, the backbone nitrogen atom and its bound hydrogen atom of Leu-19/Phe-26, and the dummy atom between the nitrogen atom and the $\alpha$ carboxyl atom of Leu-19/Phe-26 were treated quantum mechanically.

Averaged structures were generated by the tool developed by Fischer and Kandt (27). Thereby an iterative scheme of calculating an average conformation and re-aligning the trajectory to that average structure to compute a new average structure was performed. This procedure was repeated until the average structure stopped changing and the root mean square deviation

![FIGURE 3.](image_url) Comparison of the FTIR difference spectra of the hydrolysis reaction $\text{a}_{\text{hyd}}$ of Ras (blue) and Ran-RanBD1 (green, scaled by factor 1.7). The main differences occur in the $\nu(P-O_2)$-vibrational modes, the remaining part of the phosphate region is very similar.

![FIGURE 4.](image_url) Mg$^{2+}$ positions in small GTPases. Comparison of the Mg$^{2+}$ coordination in the x-ray structures with bound GTP or GTP analogue of (a) H-Ras in complex with other proteins (averaged resolution over 14 structures: 2.4 ± 0.4 Å), (b) Ran in complex with other proteins (averaged resolution: 2.7 ± 0.5 Å), (c) Rab in complex with other proteins (averaged resolution over 21 structures: 2.3 ± 0.5 Å), (d) uncomplexed H-Ras (averaged resolution over 67 structures: 1.8 ± 0.4 Å), and (e) uncomplexed Rab (averaged resolution over 39 structures: 1.9 ± 0.4 Å). The used x-ray structures in this figure can be found in supplemental Section S3.
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was below 0.001 Å. Then the structure with the closest root mean square deviation to the generated average structure was extracted from the trajectory as the run-average structure. The calculation of the B-factors from the MD simulations is described in supplemental Table S2.

Caged Compounds—The P3–1-(2-nitrophenyl)-ethylester of GTP (NPE-GTP) was synthesized in two steps from 2-nitrophenylacetylhydrazone and GTP with subsequent purification via anion exchange chromatography according to a procedure by Walker and Trentham (28). The P3-para-hydroxyphenacyl ester of GTP (pHP-GTP) was synthesized by coupling GDP and PHP-caged P. The latter was obtained in five steps from para-hydroxyacetophenone and dibenzylphosphate and according to a procedure from Park and Givens (29).

Proteins (Cloning, Overexpression, and Preparation)—Point mutations in Ras and Ran were obtained by site-directed mutagenesis using the overlap extension PCR method. Wild-type and A18T H-Ras(1–166) were prepared with Escherichia coli (CK600K) using the ptac-expression system as described (30). For Ran wild-type, RanT25A, and RanY39A(1–216) the pET3d-expression system with E. coli (BL21(DE3)) was used (8). Human RanBP1(1–201) and NF1–333 were both overexpressed as a GST (glutathione S-transferase) fusion protein using the plasmid pGEX-4T-1mod and E. coli (BL21(DE3)). The pGEX-4T-1mod plasmid has an additional tobacco etch virus protease recognition site. Purification was done by GSH affinity chromatography, on column digestion with tobacco etch virus protease, and subsequent gel filtration. Overexpression and purification of Rna1p from Schizosaccharomyces pombe (1–386) was done as previously described (31). RanBD1(1155–1321) (first Ran-binding domain of human RanBP2) was prepared as described by Vetter et al. (14).

Nucleotide Exchange and Manganese Loading with Different GTPases—For FTIR measurements Ras, Ran, and their mutants were loaded with caged GTP according to John et al. (32). A minimum 1 mg of GTase was incubated with a 1.5 M excess of pH-GTP and 1 unit of alkaline phosphatase in reaction buffer (50 mM Tris-HCl, 10 mM ZnSO₄, 200 mM (NH₄)₂SO₄, 1 mM DTT, pH 7.5). The reactions were observed via HPLC. For Ran, a 1:1 complex of Ran-RanBP1 was used.

The exchange to Ran-NPE-GTP-Mn²⁺ was a two-step process. First, nucleotide-free Ran:RanBP1 was produced according to John et al. (32) and second, following a 10-min EDTA treatment (1 mM), the GTase was loaded with 20 mM Mn²⁺ and 1.5 M excess NPE-GTP. The use of NPE-GTP instead of pH-GTP was necessary because pH-GTP is not stable in the presence of Mn²⁺ ions. After nucleotide exchange all samples were subsequently applied to a NAP-5 gel filtration column to change to low salt buffer for FTIR sample preparation (Ras: 500 μM Hesper buffer, 50 μM MgCl₂, 50 μM DTT, pH 7.4; Ran: 1 mM Hesper buffer, 100 μM MgCl₂/MnCl₂, 100 μM DTT, pH 7.6) and removal of excess nucleotide.

FTIR Spectroscopy and Sample Preparation—Samples were prepared between CaF₂ windows as described (33) and measured on a Bruker Vertex 80V spectrometer. Ran measurements were done with 5 mM concentration of the GTase in buffer composed of 200 mM Hepes buffer, pH 7.5, 20 mM MgCl₂, and 20 mM DTT. Intrinsic reactions with Ran, Ras, and mutants were carried out at 283 K, and manganese measurements with Ran at 278 K. Photolysis of the caged compounds and data acquisition was performed as described (34). The data were analyzed between 1800 and 950 cm⁻¹ with a global fit method (35). In this analysis, the absorbance changes ΔA during the hydrolysis reaction were analyzed by fitting the data to a one exponential function with an apparent rate constant k̄hydrod and the amplitude ahydrod(t) (Equation 1).

\[ ΔA(v, t) = a_{ph}(v) + a_{hydrod}(v)(1 - e^{-k_{hydrod}t}) \] (Eq. 1)

The photolysis spectrum a_{ph}(v) is a difference spectrum comparing the state before triggering with the state after flashing, extrapolated to t = 0. In the figures, the disappearing bands are shown face downwards and the appearing bands face upwards. The absolute absorbance differences depend on the exact sample thickness, the laser intensity, and the amount of caged GTP after nucleotide exchange. Thus, for the comparison of two spectra, one can be scaled to match the absolute absorbance difference of the other.

Protein Preparation for X-ray Crystallography—Exchange of bound GDP to Gpp(NH)p, a nonhydrolysable GTP analogue, on RanY39A was performed by incubating 1 μmol of each RanY39A and RanBD1 with a 2.5 excess of Gpp(NH)p and 20 units of alkaline phosphatase in a volume of 2.7 ml for 20 h at room temperature in 50 mM Tris buffer, pH 7.6, 200 mM (NH₄)₂SO₄, 5 mM dithiothreitol. The formed complex was purified on a prepacked Superdex 75 pg 16/60 column (GE Healthcare) used at an AKTAPrime System (GE Healthcare) in 20 mM Tris buffer, pH 7.4, 2 mM MgCl₂, and 2 mM DTT. Peak fractions were collected; purity was checked by SDS-PAGE, pooled, and concentrated (Amicon Ultra-15 centrifugal filter unit with molecular mass cutoff 30,000 Da, Millipore) up to 24.4 g/liter, shock frozen, and stored at −80 °C.

Protein Crystallization and Structure Solution—RanY39A (1–216)-Gpp(NH)p-RanBD1 was prepared at 24.4 mg/ml in 20 mM Tris, pH 7.4, 2 mM MgCl₂, and 2 mM DTT. Hesper drops were set up at 20 °C by mixing 1.5 μl of protein with 1.5 μl of reservoir consisting of 1.8 M ammonium sulfate, 2.5% PEG 1500, and 100 mM Hesper, pH 7.5. Crystals were flash frozen in 1.8 M ammonium sulfate, 2% PEG 1000, 100 mM Hesper, pH 7.5, and 30% glycerol. Data were collected at a wavelength of 1.039 Å using a Pilatus 6M detector. The crystals in space group C222₁ diffracted to 3.2 Å, contained 2 complexes per asymmetric unit and were solved using PHASER with the RanΔ191-BeF structure as a template. Refinement in PHENIX with local NCS and secondary structure constraints as well as RanΔ191-BeF as a reference structure (i.e. dihedral angles are constrained to the reference structure to get better geometries at this low resolution) converged to a final R and Rfree of 22.2 and 25.5%, respectively, with 0.76% Ramachandran outliers. Comparison of the model refined with and without using constraints from the reference structure showed no major differences except for better geometries in the former case.

RanΔ191GDP-BeF₃-RanBD1 was prepared in buffer “A” (20 mM KH₂PO₄, pH 7.4, 2 mM MgCl₂, 1 mM DTT, 5% (w/v) sucrose)
Results and Discussion

Influence of the Magnesium Positions on the Phosphate Vibrations—As introduced above, one reason for the well known difference of the phosphate absorptions in the FTIR spectra of GTP bound to Ras compared with Ran-RanBD1 (see Ref. 11 and Fig. 3) could be a different Mg\(^{2+}\) coordination of the triphosphate according to the corresponding x-ray structures (PDB codes 1QRA (Ras) and 1RRP (Ran-RanBD1)) called βγ-form and αβγ-form in the following. To investigate whether this might be the cause of the band shifts, we performed biomolecular simulations of H-Ras-GTP as well as of Ran-GTP-RanBD1 to be able to calculate the theoretical infrared spectra for comparison with the experimentally determined phosphate vibrations. The high resolution x-ray structure 1QRA (βγ-form) was employed as a starting structure for H-Ras, for Ran-GTP-RanBD1 the complex of Ran-Gpp(NH)p with RanBD1 (x-ray structure 1RRP) was used (αβγ-form). Additionally, we simulated both H-Ras in a Ran-like αβγ-form, and Ran with a Ras-like βγ-form. In PDB code 1QRA a crystal contact leads to an artificially "flipped out" conformation of Tyr-32. As discussed previously (12), this tyrosine moves in our simulations into the canonical position close to the γ-phosphate that is found in the majority of Ras crystal structures. This structure corresponding to the state 2 observed in NMR spectroscopy (36) with the canonical position of Tyr-32 was used as starting structure for all simulations. All four systems were stable during 50-ns simulations. The four structures, averaged over the equilibrated last 25 ns, are shown in Fig. 5.

Next, we did QM/MM simulations of six snapshots of each system and calculated the theoretical infrared spectra. To obtain reliable spectra, we treated a large part of the active site on the starting structure. Oxygen atoms are red, nitrogen atoms are blue, phosphate atoms are orange, hydrogen atoms are light gray, and the magnesium ion is purple.
binding partners of the α-phosphate are treated quantum mechanically and the calculated energy of the α-phosphate vibration should be very accurate. The results of these calculations are summarized in Fig. 6.

The calculated frequencies for $\nu_{(P\_O)}$ of Ras are 1266 cm$^{-1}$ for the $\beta\gamma$-form and 1237 cm$^{-1}$ for the $\alpha\beta\gamma$-form, respectively. As expected, the coordination of the α-phosphate by the Mg$^{2+}$ leads to a strong red shift of the vibration. Although the value of the $\beta\gamma$-form is very close to the experimental value of 1266 cm$^{-1}$, the $\alpha\beta\gamma$-form deviates by 25 cm$^{-1}$. For Ran-RanBD1 the corresponding vibration is already red shifted compared with Ras for the $\beta\gamma$-form and calculated to be at 1241 cm$^{-1}$. For the $\alpha\beta\gamma$-form a further red shift to 1229 cm$^{-1}$ is calculated. Thus, the value for the $\beta\gamma$-form coincides with the experimental value, whereas the value of the $\alpha\beta\gamma$-form deviates by 12 cm$^{-1}$. Accordingly, the calculations suggest the $\beta\gamma$-form for both proteins, Ras and Ran-RanBD1.

To obtain further experimental evidence for a similar Mg$^{2+}$ coordination of Ras and Ran-RanBD1 in solution we performed an exchange of Mg$^{2+}$ to Mn$^{2+}$. Mn$^{2+}$ is known to coordinate GTP in Ras in the same manner as Mg$^{2+}$ and selective shifts of the vibrations of coordinated phosphate groups can be expected (37). The spectra obtained via photolysis of caged GTP are shown in Fig. 7 and the shifts are summarized in Table 1. The observed shifts are comparable for Ras and Ran-RanBD1, again suggesting a similar coordination of the metal ion. In both cases no significant shift of the $\nu_{(P\_O)}$ vibration is observed, indicating no direct interaction of the Mn$^{2+}$ with the α-phosphate. According to our experience, the α-phosphate should clearly show a shift upon changes in its environment, whereas the effect on the γ-phosphate is usually less clear. Furthermore, the γ-phosphate environment in the MD simulations was not treated quantum mechanically so that the predictions would be not as accurate as for the α-phosphate. In summary, theory and experiment consistently suggest Mg$^{2+}$ coordination to the $\beta$- and $\gamma$-phosphates ($\beta\gamma$-form) for both proteins, Ran and Ras.

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**Table 1**

**Band shifts in Ras and Ran**

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*Analysis of the Catalytic Site of the GTPase Ran*

FIGURE 6. Comparison of the asymmetrical α-phosphate stretching modes of GTP bound to Ras or Ran, respectively. The experimental values of the hydrolysis difference spectra of Ras and Ran are compared with the calculated ones with both $\beta\gamma$- and $\alpha\beta\gamma$-coordination of the Mg$^{2+}$. The corresponding IR difference spectra are shown in Fig. 7.

FIGURE 7. Scaled FTIR difference spectra of (a) photolysis $\sigma_{ph}$ and (b) hydrolysis $\sigma_{hyd}$ of RasWT-GTP-Mg$^{2+}$ (light blue), RasWT-GTP-Mn$^{2+}$ (dark blue), RanWT-GTP-Mg$^{2+}$-RanBD1 (light green), and RanWT-GTP-Mn$^{2+}$-RanBD1 (dark green). The observed band shifts are summarized in Table 1.
shifted by 1.9 Å (Fig. 8). In our higher resolution structure, the magnesium is almost exactly in the same position as in the Ras structures that usually have a very good resolution and quality (e.g. PDB codes 5P21 and 2RGE). In 1RRP, the magnesium ion occupies a position where other structures have a water molecule that coordinates the magnesium ion. Indeed, analysis of all available structures of Ran (Fig. 4 and Ras (Fig. 4, a and d) reveals that the magnesium ions are found spread out along a line that connects two of the axial water molecules and the Mg$^{2+}$. It appears that, especially in lower quality structures, the Mg ion often accidentally wanders off into the water positions during refinement. This is nicely seen in Fig. 4 where high resolution x-ray structure of uncomplexed GTPases (Fig. 4, second row) reveal a clear By-form, whereas in the usually lower resolved x-ray structures of complexed GTPases (Fig. 4, upper row) the Mg$^{2+}$ coordination varies. In summary, this would suggest that the By-form is the correct magnesium position for Ras and Ran, and the aBy-form would be a crystallographic artifact due to low resolution data.

In our new Ran-RanBD1 structure, the beryllium fluoride superimposes nicely with the γ-phosphate oxygens of the Gpp(NH)p in 1RRP, and is only slightly rotated around an axis from the beryllium to the closest β-oxygen of the GDP (Fig. 8). The switch regions are, as expected, very similar to 1RRP, just the tip of switch II with Gln-69 is slightly more retracted from the γ-phosphate position, perhaps due to the slightly bulkier beryllium fluoride. As usual, switch II shows slightly elevated temperature factors. However, Gln-69 has a well defined electron density with distances of 5.4 and 5.73 Å between the GluCβ atom and γ-phosphate.

Efforts to obtain a Ran-GDP-BeF$_3$ structure without RanBD1 were not successful: two structures of Ran wild-type in different space groups were solved (supplementary Table S3) that grew in a GDP-BeF$_3$ containing buffer, but contained Ran only in the canonical GDP form, no trace of density at the γ-phosphate position that could correspond to BeF$_3$ was visible, corroborating the hypothesis that RanBD1 stabilizes the GTP conformation of Ran (14) and enables binding of BeF$_3$.

Influence of Thr-25 of Ran on the α-Phosphate Vibrations—If the Mg$^{2+}$ coordination is not the cause for the spectral difference between Ras and Ran-RanBD1, what is the reason for the red shift of the α-phosphate vibration in Ran-RanBD1 relative to Ras, found in both experiment and theory? One explanation could be the additional hydrogen bond of the Thr-25 side chain to an α-phosphate oxygen in Ran. This hydrogen bond is not present in Ras because an alanine is in the corresponding position.

To prove this idea, we performed the FTIR spectroscopic measurements of the corresponding mutants H-RasA18T with an additional hydrogen bond to the α-phosphate and RanT25A that lacks this bond, respectively. The mutation A18T in N-Ras has been found in tumors with very good prognosis (40). Demunter and co-workers (40) measured the $K_{D}$ of the nucleotide for this mutant and found it unchanged compared with wild-type (supplementary Fig. S2 shows infrared difference spectra of Ras wild-type compared with RasA18T obtained via photolysis of caged GTP (cgGTP) bound as the nucleotide). In the photolysis spectrum bands facing downward (disappearing bands) are due to Ras-cgGTP and bands facing upward (appearing bands) are due to Ras-GTP. In the hydrolysis spectrum bands facing downward are due to Ras-GTP and the bands facing upward are due to Ras-GDP + P$_i$. All bands that do not change during the reaction are cancelled out in these difference spectra. It can be seen that the spectra of wild-type and mutant are very similar.

The corresponding experiments for Ran wild-type and RanT25A are shown in supplementary Fig. S3. In all experiments, the almost unchanged spectra in the amide region between 1300 and 1700 cm$^{-1}$ indicate that the observed processes are not influenced by the mutations. Only in the region of the phosphate absorptions are some changes observed. This region is shown enlarged in Fig. 9. The asymmetric stretching vibrations of the β-phosphate ($v_{1} (P_{\alpha}O_{\beta})$) at 1216 cm$^{-1}$ and γ-phosphate ($v_{2} (P_{\alpha}O_{\gamma})$) at 1143 cm$^{-1}$, observed as positive bands in the photolysis spectrum, are unchanged. At the same time the asymmetric stretching vibration of the α-phosphate ($v_{3} (P_{\alpha}O_{\gamma})$, observed as a negative band in the hydrolysis spectrum, is downshifted by 21 cm$^{-1}$ upon the mutation A18T in Ras. Thus, this band is now almost in the position of the $v_{1} (P_{\alpha}O_{\beta})$ band of Ran. Conversely, the $v_{2} (P_{\alpha}O_{\gamma})$ band of Ran shifts upwards by 19 cm$^{-1}$ upon T25A mutation, close to its position in Ras. Similar to Ras, no significant changes are found for β- and γ-phosphate vibrations upon mutation of T25A in Ran. In summary, the phosphate bands of RasA18T resemble very precisely the corresponding spectra of Ran wild-type. Vice versa, the RanT25A spectra are very similar to Ras wild-type. Thus, the FTIR spectroscopic experiment reveals that the different α-phosphate band positions in Ran compared with Ras are not due to a different coordination of Mg$^{2+}$ but due to the additional hydrogen bond of Thr-25 of Ran that is substituted by an alanine in Ras wild-type.

Complementary, we performed MD simulations on the corresponding mutated proteins, i.e. RanT25A-GTP-RanBD1 and...
be explained by the additional hydrogen bond of the hydroxyl group of Thr-25.

Structural Details of the Triphosphate: Distance of $\beta$- and $\gamma$-Phosphate—If the differences in the hydrolysis rate of H-Ras and Ran-RanBD1 cannot be explained correctly by a different Mg$^{2+}$ coordination, what else can be the reason? Now that we have dynamic computational models of Ras-GTP-Mg$^{2+}$ and Ran-GTP-Mg$^{2+}$-RanBD1 with atomic resolution, which are validated by the agreement of experimental and calculated infrared spectra, we can examine them with regard to structural parameters far beyond the resolution of x-ray crystallography and with GTP instead of being limited to analogues like Gpp(NH)p or GDP-BeF$_3$. As shown above this high resolution is essential to elucidate catalytic effects. Fig. 10 shows the small structural differences between GTP in water, bound to Ran and bound to Ras. Interestingly, the distance of the $\gamma$-phosphate to the $\beta$-phosphate increases from 2.99 Å in water to 3.00 Å in Ran but to 3.03 Å in Ras. The faster the reaction the larger is this distance. Similarly the P$_{\beta}$OP$_3$ angle increases. This could be one cause for the slower GTPase reaction of Ran compared with Ras.

Structure of RanY39A-RanBD1—Another observation hints to the influence of the side chain of Tyr-39. The Y39A mutant of Ran was found to catalyze GTP hydrolysis about 2 orders of magnitude faster compared with Ras wild-type, whereas in Ras the corresponding Y32A mutation has almost no influence on GTP hydrolysis compared with wild-type (Fig. 2) (11). We therefore tried to solve the structure of RanY39A to elucidate the reason for the faster hydrolysis. As for RanWT, we first tried to obtain a structure of RanY39A alone (without binding partner) in complex with GDP-BeF$_3$ and solved two 1.75-Å crystal structures in different space groups (supplemental Table S3), but only GDP was observed in the active site, similar to wild-type Ran where a Ran binding domain is needed to force Ran-GDP into the GTP conformation and thus enables BeF$_3$ binding. The region around the mutated Tyr-39 did not show any significant changes relative to wild-type, and the same is true for the remainder of the protein.

In contrast, we could crystallize the RanY39A-Gpp(NH)p-RanBD1 complex (RanY39A-RanBD1 in the following) and solve the structure with 3.2-Å resolution. The magnesium ions are located in a position very close to the Ran-RanBD1 and Ras positions, although at a resolution of 3.2 Å this is not very reliable as mentioned above.

Close to the active site, switch II shows a relatively large deviation of residues 70–75 relative to 1RRP and Ran-RanBD1. This is most likely due to crystal packing because a neighboring molecule is so close that it “flattens” the switch II helix and pushes Phe-72 and Gln-69 slightly forward in direction of the axis of the switch II helix. In contrast, the switch I region is almost completely unaltered compared with Ran wild-type except for the missing side chain of Tyr-39. As in the original Ran-RanBD1 structure (PDB code 1RRP) and as well as in our Ran-RanBD1 structure, the temperature factors for the switch II region are elevated, in the case of RanY39A-RanBD1 more than in the other two structures. Gln-69 has well defined electron
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The experimental value is compared to the calculated ones with both $\beta\gamma$- and $\alpha\beta\gamma$-form. The corresponding IR difference spectra are shown in Fig. 5.

TABLE 2
Comparison of the shifts of the asymmetrical phosphate stretching modes of the $\alpha$- ($\nu_\alpha(P_\alpha O_\alpha)$) and $\beta$-phosphate ($\nu_\beta(P_\beta O_\beta)$) group of GTP bound to Ras upon A18T mutation and Ran upon T25A mutation, respectively.

<table>
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<tr>
<th>Mg$^{2+}$ coordination</th>
<th>RasA18T shift</th>
<th>RanA18T shift</th>
<th>RasT25A shift</th>
<th>RanT25A shift</th>
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Figure 10. Structural changes of the triphosphate in water, bound to Ran, and Ras. The educt states are destabilized by the elongation of the distance between the $\gamma$- and $\beta$-phosphorous atom. The educt state in Ran-RanBD1 is less destabilized than in Ras so the energy barrier for hydrolysis in Ran is higher than in Ras. The values are the average over six 2.5-ps QM/MM simulations. Changes are exaggerated for clarity. The triphosphate structures in water and Ras are obtained by Rudack et al. (12).

density in both RanY39A chains with distances of 6.07 and 6.22 Å from the C$\delta$ atom of Gln-69 to the $\gamma$-phosphate. Further details are given in the supplemental Section S1.

In summary, there are no significant changes in the active site that could directly explain the different reaction rates of Ran wild-type and Y39A. This prompted us to investigate the possibility that the wild-type of Ran has a slower hydrolysis rate than the mutant because the side chain of Tyr-39 blocks access of the catalytic glutamine to the $\gamma$-phosphate as described in the following.

The role of Tyr-39 and Gln-69—The side chain of Tyr-39 is found in a position close to the $\gamma$-phosphate in all known Ran-GTP complex structures. However, structures of related GTPases show a greater variability of the tyrosine orientation, ranging from positions relatively similar to Ran (called “closed” in the following) to positions where the tyrosine side chain is far away from the $\gamma$-phosphate (“away” position). Interestingly, there seems to be a tendency that in small GTP-binding proteins with a fast GTP hydrolysis this tyrosine is preferentially in the away position according to x-ray and NMR data, and the mutation to, e.g. alanine, does not have a large effect on the hydrolysis rate (e.g. Cdc42 (42, 43)), whereas in GTPases with a slow hydrolysis the tyrosine is close to the $\gamma$-phosphate (e.g. Rab6a, which is approximately 4 times slower than Ran-GTP and 8 times slower than Ran-GTP-RanBD1) (44). Fig. 11a shows the side chains of Gln-69 and Tyr-39 in Ran-RanBD1 (PDB code 1RRP) compared with the structure of RanY39A-RanBD1 solved in this work. In the available x-ray structures of Ran wild-type, including the Ran-GDP-BeF$_3$-RanBD1 structure solved here, Tyr-39 is contacting the $\gamma$-phosphate (Fig. 8). FTIR spectroscopic measurements on RanWT-GTP-RanBD1 and RanY39A-GTP-RanBD1 confirm a direct hydrogen bond between Tyr-39 and the $\gamma$-phosphate, which can be observed by the blue shift of the $\gamma$-phosphate vibration upon Y39A mutation by 12 cm$^{-1}$ (Fig. 11b). This shift is of a similar magnitude as the one found for removing one hydrogen bond from the $\alpha$-phosphate upon T25A mutation as discussed above. How does Tyr-39 hamper GTP hydrolysis? A possible explanation is the steric hindrance depicted in Fig. 11c. It is known that Gln-61(Ras)/Gln-69(Ran) is important for the hydrolysis reaction because it holds the nucleophilic water molecule in the attacking position (45, 46) and mutations of this residues are often found in tumors (47). The tyrosine side chain intrudes between the Gln and $\gamma$-phosphate. The distance of the side chain oxygen of Tyr-39 from a line between the GlnC$\delta$ and $\gamma$-phosphor atom can be taken as a measure for the steric hindrance. This distance is 2.6 Å in Ran, but 3.8 Å in Ras, which could also explain why in Ras this effect is not seen and, accordingly, the mutation Y32A has no significant effect on the hydrolysis rate (Fig. 2). As expected, this steric hindrance is removed in the Ran-RanGAP complex (46) where the corresponding distance is 3.3 Å, corroborating the idea that the tyrosine side chain interferes with the hydrolysis.

MD Simulations of WT and Tyr Mutants—Complementary to the analysis of the crystal structures, we analyzed our MD simulations regarding the role of the discussed Tyr and Gln residues. The distance of the side chain oxygen of Tyr-39 from a line between the GlnC$\delta$ and $\gamma$-phosphor atom averaged over the simulation trajectory is 2.2 Å for Ran and 3.6 Å for Ras, in agreement with the x-ray data (Fig. 11c), confirming the steric hindrance. Furthermore, the simulation of RanWT-RanBD1 exhibits a stable positioning of Tyr-39 at the $\gamma$-phosphate during a 50-ns simulation (Fig. 12). Gln-69 is trapped in an unreactive conformation by a direct hydrogen bond to Tyr-39. In contrast, the analogous simulation of Ras reveals a fluctuating position of the Gln-61 of Ras with a larger calculated B-factor. Thus Gln-61 can acquire the reactive conformation from time to time. The different flexibility of the Gln side chain is also seen in Fig. 11d, which shows the Gln color coded by the calculated B-factors. The Gln side chain of the slow hydrolyzing Ran wild-type has a very low B-factor, whereas the faster hydrolyzing RanY39A has a much higher B-factor. In the case of Ras, the Gln side chain is already flexible in Ras wild-type and unchanged in RasY32A (supplemental Table S2), in line with a rate more comparable with RanY39A for both Ras wild-type and Y32A mutant. A similar mechanism might
Analysis of the Catalytic Site of the GTPase Ran

**Conclusion**

By a combination of theoretical IR spectroscopy and time-resolved FTIR spectroscopic measurements we have shown that Mg\(^{2+}\) is βγ coordinated in both proteins, Ras and Ran. An in-depth analysis of existing x-ray structures together with the new x-ray structures reported here confirm a clear βγ-coordination as well. Our structure of the Ran-GDP-BeF\(_3\)-RanBD1 complex has an improved resolution compared with previous structures and thus allows a more accurate view of the nucleotide binding site. The red shift of the asymmetric \(\alpha\)-phosphate vibration (ν\(_{\alpha}(P_O\alpha)\)) was shown to be due to a hydrogen bond between Thr-25 and the \(\alpha\)-phosphate oxygens in Ran that do not exist in Ras, as proven by the agreement of theoretical and experimental mutagenesis studies. Thus, the differences in the hydrolysis rate between Ras and Ran cannot be explained by a difference in the Mg\(^{2+}\) coordination. We therefore suggested that Tyr-39 of Ran interfered with the catalytically favored position of Gln-69 relative to the \(\gamma\)-phosphate in Ran via steric hindrance, whereas in Ras the catalytic competent structure can be obtained more frequently by the more flexible and less hindered Gln-61 side chain because the Tyr-32 side chain in Ras is positioned slightly different. In agreement, the hydrolysis

also be present in various related slow Rab GTPases. Especially the slow hydrolyzing Rab6a has a tyrosine in a similar position as Tyr-39 in Ran (44).
reaction of the Y39A mutant of Ran is accelerated by 2 orders of magnitude, whereas the analogue Y32A mutant of Ras is not accelerated (Fig. 2).

In addition, we present a proof of principle for the high sensitivity of the integration of theoretical and experimental IR spectroscopy. Here, we have predicted IR spectroscopic shifts of mutations, which afterward were verified by FTIR spectroscopic measurements. The difference of one hydrogen bond was resolved by both, theory and experiment. The integration of x-ray structure analysis, QM/MM simulations, and IR spectroscopy is a powerful tool to analyze the structure and dynamics of the catalytic center of a protein with atomic resolution. Only such atomic resolution provides the basis to understand protein catalysis.


Acknowledgments—We thank Dr. Frederick Großerüschkamp and Sara Stephan for the RasMn2+ and RasY32A measurements, Claus Küpper for performing RanWT-RanBD1 MD simulations, and Stefan Tennigkeit for help in the purification of the Ran and Ras mutants. We acknowledge Dr. Jürgen Schlitter for fruitful discussions. X-ray data collection was performed at the Swiss Light Source, beamline X10SA, Paul Scherrer Institute, Villigen, Switzerland. We thank the beamline staff and Toni Meinhardt, Bernhard Loll, Andrea Rocker, Wolf Blanデンフェルト, Nils Schrader, and Sven Hennig for help with the data collection.

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Molecular Biophysics:
Catalysis of GTP Hydrolysis by Small GTPases at Atomic Detail by Integration of X-ray Crystallography, Experimental, and Theoretical IR Spectroscopy

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doi: 10.1074/jbc.M115.648071 originally published online August 13, 2015

Access the most updated version of this article at doi: 10.1074/jbc.M115.648071

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Supporting Information

Catalysis of GTP hydrolysis by small GTPases at atomic detail by integration of X-ray crystallography, experimental and theoretical IR spectroscopy

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Supporting Table S1. Simulation systems with the Mg 2+ coordination, numbers of amino acids, solute atoms, water molecules, and counter-ions. The MD simulations of Ras base on the X-ray structure 1QRA. Note that during the equilibration of the structure the Tyr32 flipped into the GTP binding niche and forms a hydrogen bond to an oxygen atom of the γ-phosphate. This is the canonical structure found in most X-ray structures. The MD simulations of Ran base on the X-ray structure 1RRP.

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Supporting Table S2: B-factors of the Gln69/Gln61 side chains of the X-ray structures of RanWT (1RRP) with RanT39A and the simulated B-Factors calculated from the 50 ns trajectories of the MD simulations of RasWT, RasY32A, RanWT and RanT39A. The trajectories are fitted to the averaged structure and then the standard deviation σ of the atomic positions in the trajectory is calculated. The B-factor is 8πσ². The values are the average of the B-factors of all side chains atoms of Gln69/Gln61. The B-factors are given in the unit Å².

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Section S1. Details of the new RanY39A·RanBD1 structure
We crystallized the RanY39A·GppNHp·RanBD1 complex (RanY39A·RanBD1 in the following) and solve the structure with 3.2 Å resolution. Here, several alterations compared to the Ran wildtype complex with RanBD1 (1RRP) were observed. While the two RanΔ191·GDP·BeF·RanBD1 complexes (Ran·RanBD1 in the following) in the asymmetric unit are very similar to each other (and thus were compatible with tight NCS constraints, see methods), 1RRP and RanY39A·RanBD1 show relative rotations of 4° and 3.3°, respectively, between the Ran and RanBD1 domains of the two complexes in the asymmetric unit. RanY39A·RanBD1 compared to 1RRP shows rotations between 3.4° and 8.9°, Ran·RanBD1 compared to 1RRP even 7.8°-11.7°. Ran·RanBD1 and RanY39A·RanBD1 are more similar with values between 2.9° and 5.7°. This means that no two rotations are similar except for the two Ran·RanBD1 complexes in the asymmetric unit, indicating that the interface is relatively variable and that interactions outside of the interface (e.g. the N-terminus of the RanBD1 domain) contribute to the interaction. In addition, RanY39A·RanBD1 exhibits a slightly different position of the Ran C-terminal helix, but with a conserved position of the start of the 211DEDDDL216 motif (in both 1RRP and RanY39A·RanBD1 the last amino acid with a well-defined electron density is Asp211). Interestingly, a disulfide bridge is formed between the cysteines 142 of two symmetry related neighboring RanBD1 molecules in both complexes of the asymmetric unit, which might contribute to the observed differences.

Another difference compared to 1RRP and Ran·RanBD1 is a shift of the N-terminus of RanBD1. In one complex in the asymmetric unit, the defined electron density ends with residue 22 with weak density for the following residues 21-15 that are in the “canonical” position. In the other complex density stops with residue 19, and the “canonical” position is blocked by a symmetry related molecule so that residues 19-24 follow a different path compared to 1RRP and Ran·RanBD1.

Six putative sulfate ions, i.e. blobs of electron density that were too strong for water, are observed in the RanY39A·RanBD1 structure as it was crystallized in 1.8 M ammonium sulfate. One binding site is formed by rearrangement of Arg140 of RanY39A (that contributes to the “basic patch” that fixes the C-terminal 211DEDDDL216 motif of Ran in the Ran·GDP form). The Arg140 side chain has moved together with the short helix that contains R140 compared to 1RRP and Ran·RanBD1. Otherwise it would clash with a neighboring molecule in the RanY39A·RanBD1 structure. The resulting conformational change pushes loop 141-144 in direction of the canonical binding site of the RanBD1 N-terminus so that it is displaced relative to 1RRP as described above. Interestingly, a very similar conformation of the region 130-144 was observed in the Ran-Importin beta complex 3EA5 (1). In general, this stretch is relatively variable when comparing all other available Ran structures. In chain C of the RanY39A·RanBD1 structure, residues 140-143 are disordered so that the N-terminus of RanBD1 (chain D) can assume its canonical place. The rearranged arginine side chain of Ran in the first RanY39A·RanBD1 complex (chains A and B) forms a binding site for a putative sulfate ion close to His105 and Pro102 of the same Ran chain. Another sulfate ion is observed close to Arg108 of RanBD1 that corresponds to the basic patch of RanBD1 that also interacts with the DEDDDL motif of the Ran C-terminal helix, confirming the preference of this region for negative charges.

References:
## Supporting Table S3: X-ray data collection and refinement statistics

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* Values in parentheses are for highest resolution shell
Section S2. Used X-Ray structures in Figure 1 c
1RRP, 1K5D, 1QBK, 1WA5, 2BKU, 2X19, 3A6P, 3ICQ, 3M1I, 3NC0, 3NC1, 3W3Z, 3ZJY, 4HAT, 4HAU, 4HAV, 4HAW, 4HAX, 4HAY, 4HAZ, 4HB0, 4HB2, 4HB3, 4HB4

Section S3. Used X-Ray structures in Figure 4
a) Complexed Ras structures:
   2uzi, 2vh5, 3ddc, 2c5l, 1k8r, 1lfd, 1he8, 4g0n, 4g3x, 1nvv, 1nvw, 1nvu, 1nvx, 4k81
b) Complexed Ran structures:
   1rrp, 1ibr, 1qbk, 1wa5, 2bku, 2x19, 3a6p, 3gjx, 3icq, 3m1i, 3nbz, 3nc0, 3nc1, 3w3z, 3zjy, 4c0q, 4gmx, 4gpt, 4hat, 4hau, 4hav, 4hax, 4hay, 4haz, 4hb0, 4hb2, 4hb3, 4hb4, 1k5d,
c) Complexed Rab structures:
   4d0l, 3tjd, 1zbz, 3bc1, 2zet, 1tu3, 3mjh, 3bbp, 3cwz, 1yhn, 1z0k, 2d7c, 2hv8, 2gzh, 4c4p, 3tso, 3qbt, 3tnf, 1z0j, 4fz

d) Uncomplexed H-Ras structures:
   5p21, 1p2s, 1p2t, 1p2u, 1p2v, 1ctq, 3tpg, 3k9n, 3k9l, 2rge, 3k8y, 3lbh, 3lb1, 3lbn, 3oiu, 3oiv, 3oiv, 3l8y, 3l8z, 3ny, 3rz, 3s0, 3s2, 3s3, 3s4, 3s5, 3r3, 3rso, 3v4f, 4dir, 4dis, 4dit, 4dlu, 4dlv, 4dlw, 4dlt, 4dlx, 4dly, 4diz, 4e8f, 4efm, 4efn, 6q21, 121p, 1qra, 421p, 1jah, 221p, 521p, 621p, 721p, 821p, 1jai, 2rrg, 1f0, 1zw6, 2rgb, 2rgc, 2rgd, 3tjd, 3tj5, 3km, 3kén, 1xcn, 1aq, 1agp
e) Uncomplexed Rab structures:
   3nkv, 3rab, 2bme, 1yu9, 1n6h, 1n6o, 1n6p, 1n6l, 1r2q, 1huq, 1z07, 1yzq, 2y8e, 2gil, 2ffq, 1191, 1vg8, 3law, 4hlw, 1yze, 2ocb, 1yzk, 1oiw, 2f9m, 1x3s, 1z08, 1yzt, 1yzu, 1yvd, 2g6b, 3e8h, 2e1w, 2fg5, 1z06, 1yzn, 1ky2, 1ek0, 1g17
Supporting Figure S1: Comparison of the averaged structures of the equilibrated last 25 ns of 50 ns molecular dynamics simulations of solvated RasWT·GTP·Mg$^{2+}$·RanBD1 (light blue carbon atoms) (a) and RasA18T·GTP·Mg$^{2+}$ (b) as well as RanWT·GTP·Mg$^{2+}$·RanBD1 (light green carbon atoms) (c) and RanT25A·GTP·Mg$^{2+}$·RanBD1 (d). In (c) and (d) the second water molecule HOHX does not exist in the X-ray structure of RanWT·RanBD1 (1RRP) and has been added referring to the Ras structure of the βγ-form. In all four 50 ns simulation trajectories of Ran and Ras the Mg$^{2+}$ is βγ-coordinated in the starting structure and remained stably bidentately coordinated by the triphosphate. Oxygen atoms are red, nitrogen atoms are blue, phosphate atoms are orange, hydrogen atoms are light grey, and the magnesium ion is purple.
Supporting Figure S2: The phosphate and amide region of the FTIR difference spectra of the photolysis reaction $a_{ph}$ (a) and hydrolysis reaction $a_{hyd}$ (b) in RasWT (blue) and RasA18T (green). The mutant is scaled by factor 3.
Supporting Figure S3: The phosphate and amide region of the FTIR difference spectra of the photolysis reaction \( \text{aph} \) (a) and hydrolysis reaction \( \text{ahyd} \) (b) in RanWT·RanBD1 (green) and RanT25A·RanBD1 (blue). The mutant is scaled by factor 2.
Supporting Figure S4: Comparison of the distances of the key residues in Ras and Ran. Given are the distances between the oxygen atom of the Tyr side chain (32 in Ras and 39 in Ran), the C\(\delta\) atom of the Gln (61 in Ras and 69 in Ran) and the phosphor atom of the \(\gamma\)-phosphate. The values are averaged over the published complexed X-ray structures of Ras and Ran compared to the averaged values of the MD simulation trajectories. All values are given in Å. The used X-Ray structures for the values in this figure and in Figure 7 c are:
complexed H-Ras structures: 2uzi, 2vh5, 3ddc, 2c5l, 1k8r, 1lfd, 1he8, 4g0n, 1nvv, 1nvw, 1nvu, 1nvx, 4k81
Complexed Ran structures: 1rrp, 1ibr, 1qbk, 1wa5, 2bku, 3a6p, 3w3z, 4gmx, 4gpt, 4hat, 4hau, 4hav, 4haw, 4hax, 4hay, 4haz, 4hb0, 4hb2, 4hb3, 4hb4, 1k5d